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Hydrogen + oxygen → water $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ B. Write a balanced chemical equation for the reaction of carbon dioxide with limewater. The formula for the active substance in limewater is on the pipet, and the products of the reaction are calcium carbonate and water. $\text{Ca}(\text{OH})_2(\text{aq}) + \text{CO}_2(\text{g}) = \text{CaCO}_3(\text{s}) + \text{H}_2\text{O}(\text{l})$ Good job with the equations.

Labpaq - 725 Words | Bartleby

Labpaq Answer Key Psychodynamic Answer Key: D Question 2 of 10 0.0/ 2.0 Points Researchers are studying the effects of nicotine on driving. The participants are divided into two groups. One group is provided with nicotine cigarettes and the other with fake cigarettes that taste and smell like real cigarettes but do not contain nicotine.

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