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Form a bond: Each electron has a charge of $1-$, and each proton has a charge of $1+$. You can calculate the charge of an

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atom by subtracting the number of electrons from the number of protons.

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Quiz: Ionic Bonds

This worksheet (and Answer Key) is a great way to assess students' prior knowledge of ionic and covalent

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bonding. It is a great pre-assessment for high school chemistry classes, and a wonderful review activity for middle and high school classes that have already learned about bonding in class.

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Ionic bond, also called electrovalent

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bond, type of linkage formed from the electrostatic attraction between oppositely charged ions in a chemical compound. Such a bond forms when the valence (outermost) electrons of one atom are transferred permanently to another atom.

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