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AnswersChapter 13 States of Matter 1 Chapter 13 States of Matter 2 Kinetic Theory as Applied to Gases Fundamental assumptions about gases. The particles in a gas are considered to be small, hard spheres with an insignificant volume. Chapter 13 States Of Matter Worksheet Page 12/29.

Chapter 13 States Of Matter Chemistry Test Answers

CHAPTER 13,States of Matter (continued) SECTION 13.3 THE NATURE OF SOLIDS (pages 396–399) This section describes the highly organized structures of solids, distinguishes between a crystal lattice and a unit cell, and explains how allotropes of an element differ. A Model for Solids (page 396) 1. Is the following sentence true or false?

Name Date Class STATES OF MATTER 13

Answers 1. Energy of motion. 2. Faster. 3. The temperature where all particle motion ceases. 4. Cool superconducting magnets. 13.6 Physical Properties and Intermolecular Forces Practice

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the answer. You are already familiar with the three common states of matter: solid, liquid, and gas. Solid objects litter the room ... 13.1 The Fluid States 300 States of Matter FIGURE 13-1The ice cube, a solid, has a definite shape. But water, a fluid, takes the shape of its container.

Chapter 13: States of Matter

small, hard Chapter 13 States Of Matter Answer Key Title: Chapter 13 States of Matter 1 Chapter 13 States of Matter 2 Kinetic Theory as Applied to Gases Fundamental assumptions about gases. The particles in a gas are considered to be small, hard spheres with an insignificant volume. Between particles in a gas there is empty space. No attractive ...

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